

Hybrid Bonding Methods Using Ultra Precision Cutting for 3D-SIC

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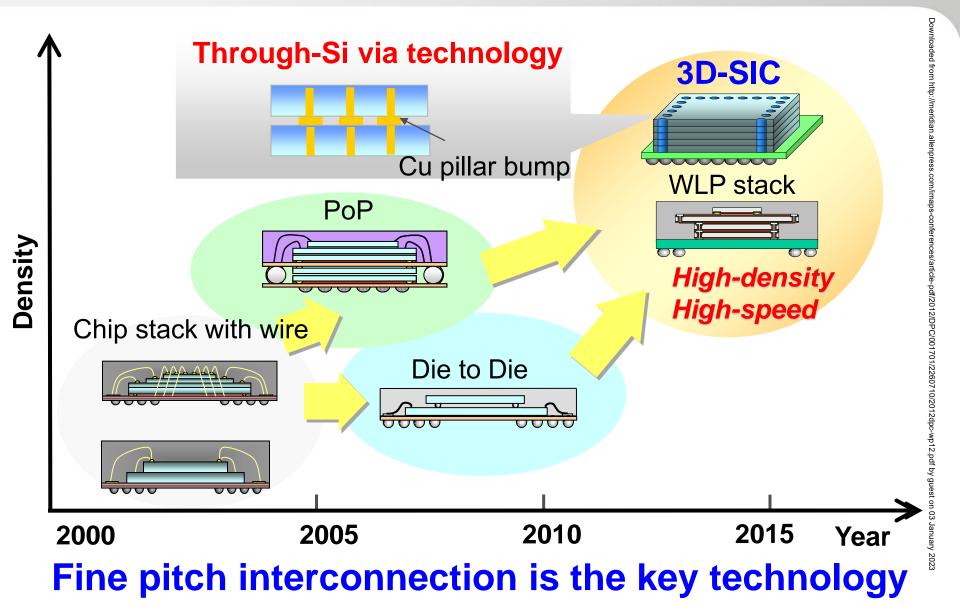
Outline



- Background and Objective
 - Demand of underfilling for 3D-SIC integration
- Hybrid Bonding Technology
 - Requirements
 - Low temperature Cu-Cu bonding technology
- Investigations
- Conclusions

System in Package Technology Trends





Fine Pitch interconnection for 3D-SIC



Bump type	Cu pillar + solder	Cu pillar + Cu pillar
Bonding method	Fusion bonding	Solid diffusion bonding by thermocompression
Structure	Si Cu pillar Si Cu pillar	s-conferences/article-po
Electrical Resistivity	11 μ Ω cm	$1.7~\mu\Omega$ cm
High current density effect	Electromigration	Stable Stable >2 μm
Available bonding pitch	> 10 μm	
Bonding temperature	217-250 °C	200-250 °C *

Cu-Cu bonding has advantages for 3D-SIC

Demand for 3D-Siehibintegration Fountain Hills, AZ USA



Further integrated 3D-SIC

• Bonding pitch <5 μm

Sensor

Analog

memor\

Device Packaging 2011 • Device gap <10 μm

Cu-Cu bonding

Reliability of stacked ICs

Device Packaging 2012

Undefilling

- Attach thinner ICs
- Protect smallerbonding electrodes

Multichip stacking process

Cu-Cu bonding and underfilling process must be realized

Interporser

Convention and vs fere Hybrid Bonding Hills, AZ USA



Hybrid bonding is advantageous for 3D-SIC

		<u>. </u>
	Conventional bonding	Hybrid bonding Translation
	Underfill	Cu bump Under Und
Underfill type	Capillary filling	Pre-applied ***
Supply area size	Individual chip size	Wafer size
Narrow gap filling	Difficult	Easy
Degasification	N/A	Potentially affected Potentially affected
Underfill infiltration into interface	N/A	Potentially occurs

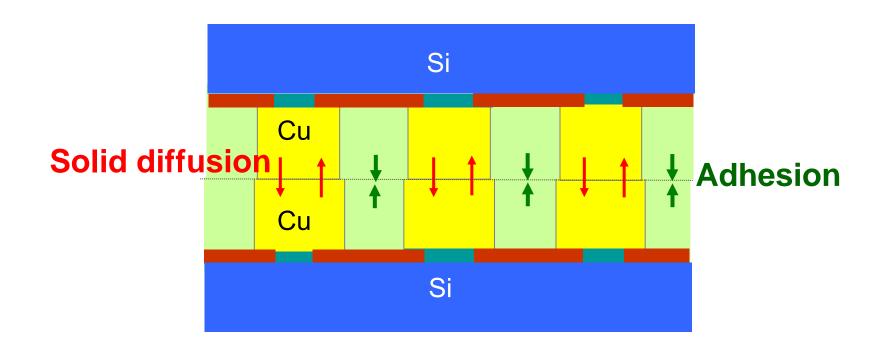
Affect to solid diffusion

Objective of hthe Present Study Fountain Hills, AZ USA



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To realize hybrid bonding with Cu-Cu solid diffusion



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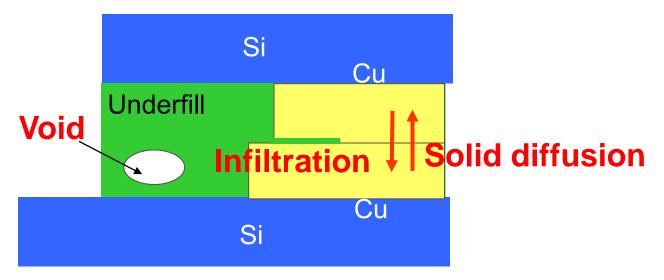
Requirements of Hybrid Bonding IIIS, AZ USA



- Low temperature Cu-Cu bonding
 - Thermostability of underfill is not too high
- Cu-Cu solid diffusion

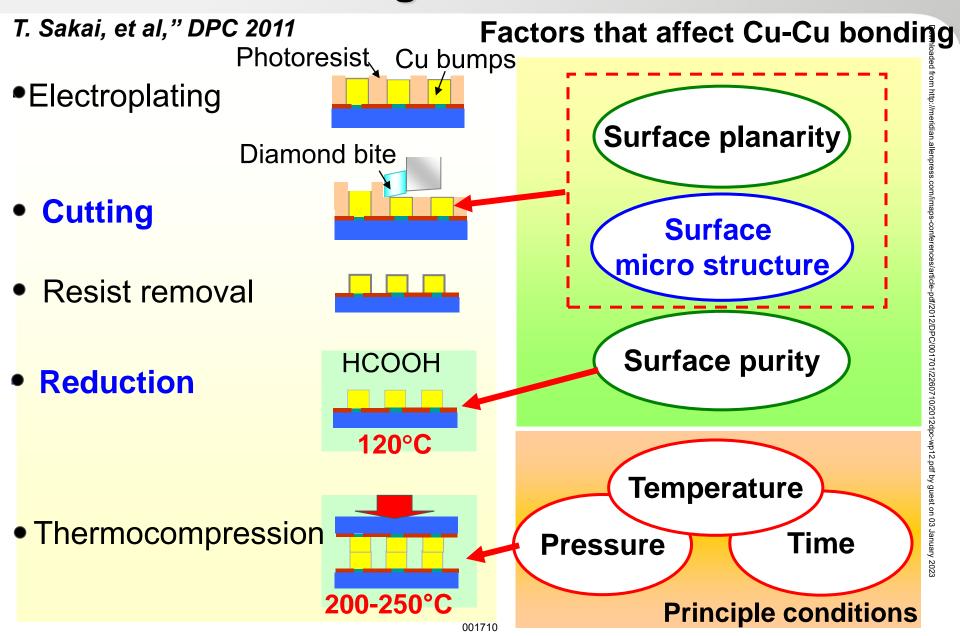
Device Packaging2011 (without underfill)

- No underfill infiltration into bonding interface
- No void formation



Our Cu-Cuap Bronding ion Process 2012 | Fountain Hills, AZ USA

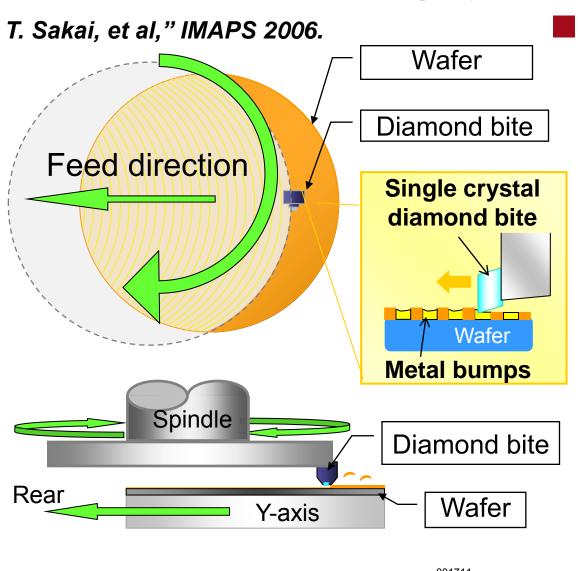




Surface Planarity Improvementuntain Hills, AZ USA



Ultra Precision Cutting by Using Diamond Bite



Feature

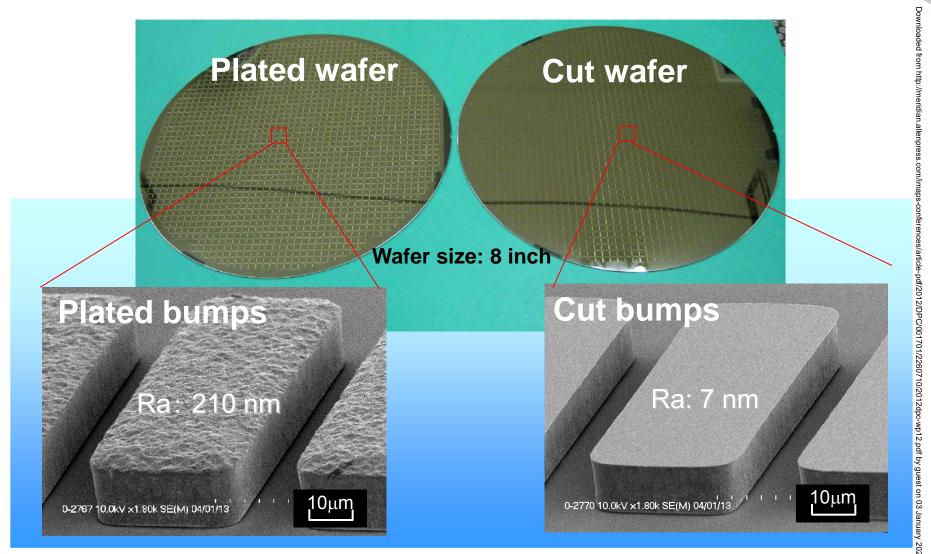
- Creep feed cutting with diamond bite
- ■Wet / dry process
- ■Various work shape



Commercialized by DISCO Corporation

Planarity of Gernatuton Beumps Packaging | March 5-8, 2012 | Fountain Hills, AZ USA

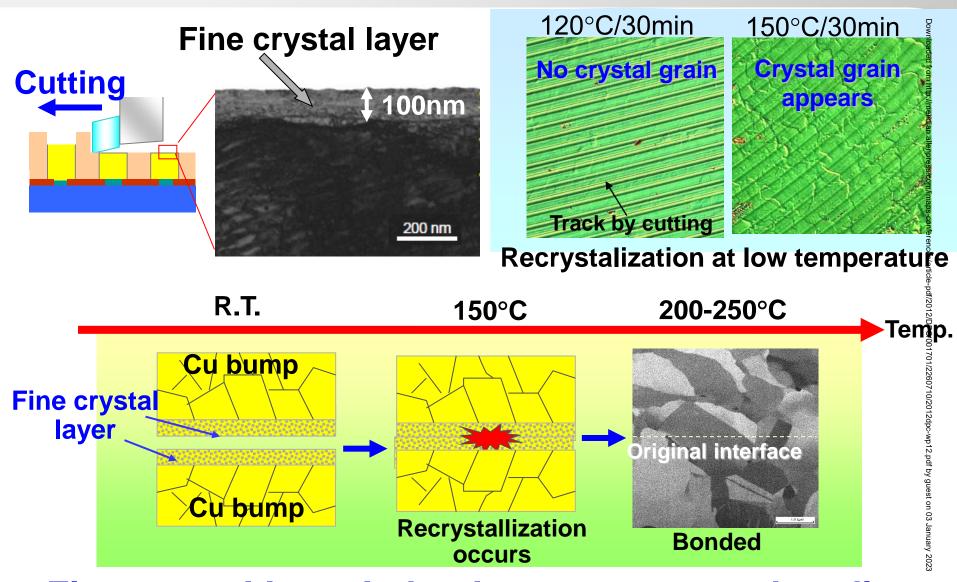




Highly flat and smooth surfaces could be obtained

Surface Micro Structure Improvement





Fine crystal layer led to low temperature bonding

Surface Purity Improvement^{8, 2012 | Fountain Hills, AZ USA}



"Reduction" is key Wet process Dry process

Sulfuric acid

Citric acid

Hydrochloric acid

Moisture absorption problem to adhesive

- Hydrogen gas
 - High temperature >300°C
- Ar ion beam
 - High vacuum condition
- Formic acid vapor
 - Low temperature >120°C

Favorable for hybrid bonding process

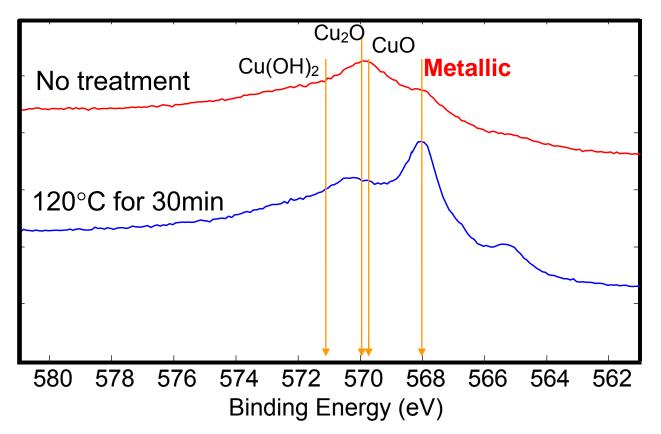
We selected formic acid vapor process

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Reduction by Committee Description Descrip



Cu LMN Auger Spectrum by XPS analysis



Metallic Cu peak was obtained in 120°C for 30min

Low temperature reduction process is possible

Outline

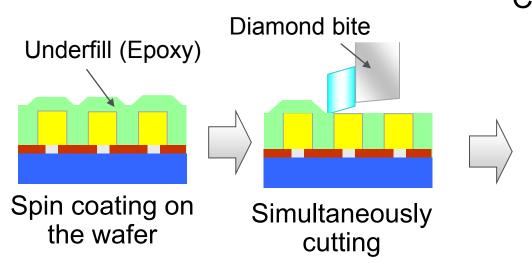


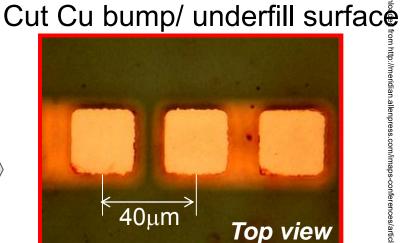
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Evaluation MAIStating Percentilition on Device Packaging | March 5-8, 2012 | Fountain Hills, AZ USA



Hybrid structure formation





Excellent cutting result

Bonding sample- Die to Die type

		Si chip	Si substrate
Chip size	mm	5×5	15×15
Chip thickness	μm	625	625
Bump size	μm	20×20	25×25
Bump height	μm	10	10



Si substrate

Investigated Bonding Process Flow



Examine how underfill is exposed by formic acid

Totally exposed by formic acid

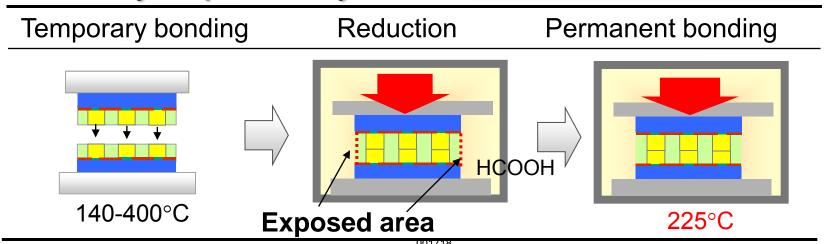
Reduction Temporary bonding Permanent bonding

Heat and pressure

140-400°C

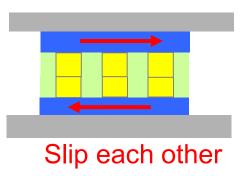
225°C

Partially exposed by formic acid





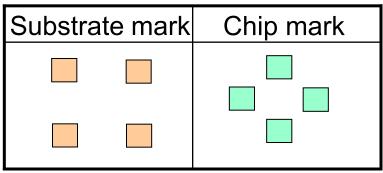
- Bonding mismatch evaluation
- Evaluation of bondability
 - Confirmation of solid diffusion
 - Void observation
- Underfill damage by formic acid
- Formic acid penetrability
- Underfill infiltration into bonding interface

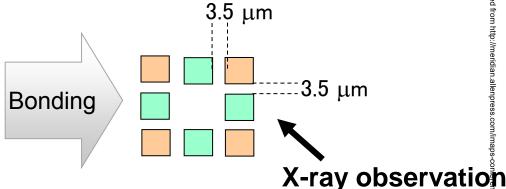


Bonding Mismatch Evaluation 1 Fountain Hills, AZ USA



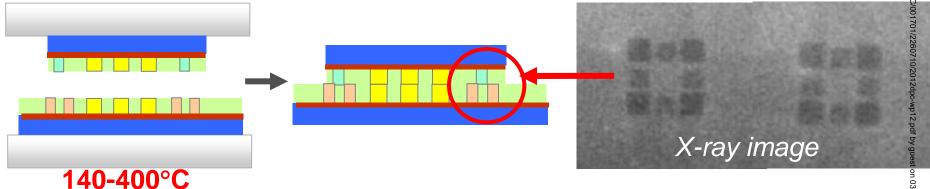
Alignment mark





X-ray images after temporary bonding

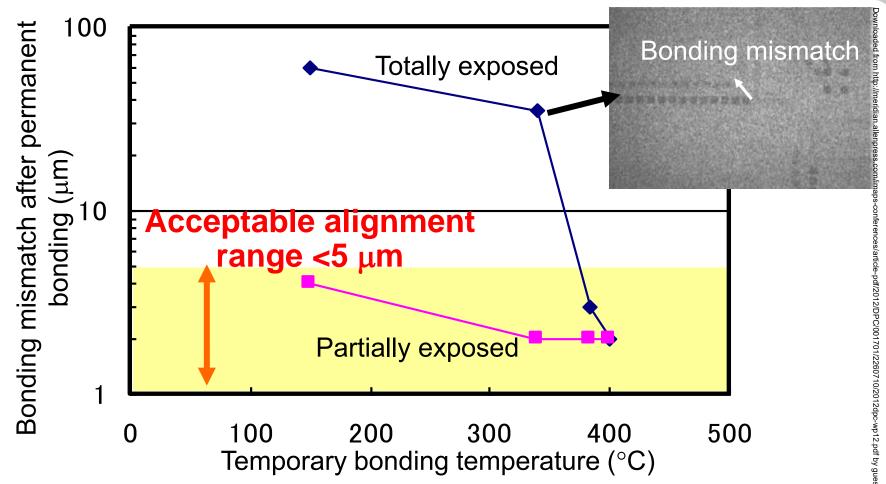
Flip chip bonder



All samples were confirmed within 2µm mismatch after temporary bonding

Bonding Mismatch after Permanent Bonding



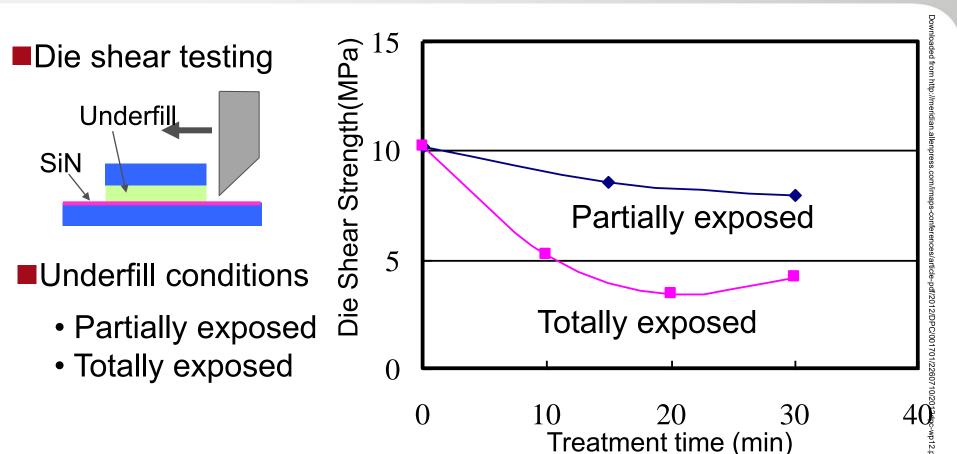


Totally exposed underfill could not sustain arranged location during bonding process

Partially exposed process shows less mismatch

Adhesion Damage by Formic Acidaz usa





Partially exposed: slightly decreases

Totally exposed: decreases below half level of the initial value

Decreases of adhesion strength by formic acid would cause mutual slipping

Bonding Reservation Strength & Exhibition on Device Packaging | March 5-8, 2012 | Fountain Hills, AZ USA



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	Totally exposed process	Partially exposed process
	Temporary bonded at 400°C	Temporary bonded at 140°C
Cross sectional images	Cu Clear interface	Solid diffusion Solid diffusion
C-SAM images	Void existed	No void

C-SAM: Constant-depth mode Scaning Acoustic Microscorpe

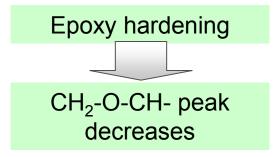
Solid diffusion and voidless bonding could be achieved in partially exposed process

Underfill Damage by Formic 5-8 A2cid HIIS, AZ USA



Microscopic FT-IR analysis

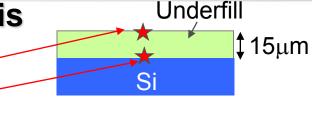
- 1. Uncured
- 2. Formic acid/ top side
- 3. Formic acid/ Si side
- 4. Completely cured

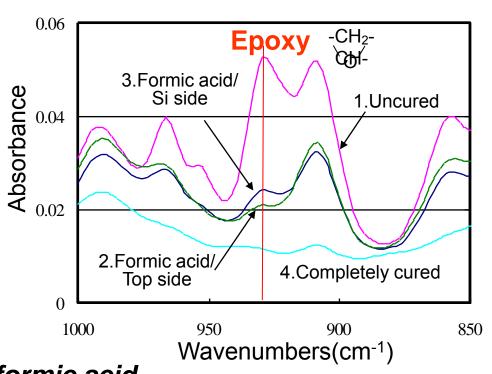


Normalized cure degree

Formic acid/ top side: 88%

Formic acid/ Si side: 70%





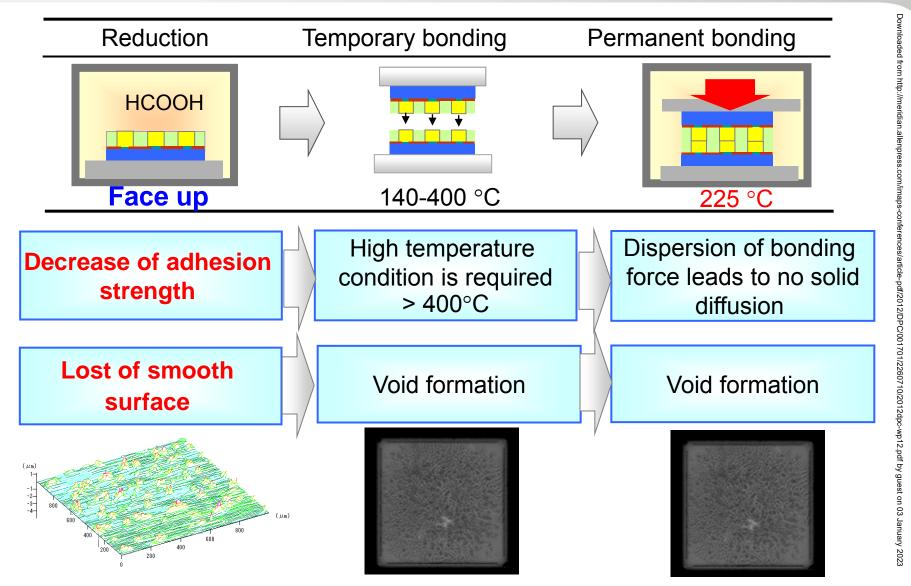
Ring-opening reaction by formic acid

 $HCOOH + R-O-CH_2-CH-CH_2 \longrightarrow R-O-CH_2-CH-CH_2 \bigcirc O-CHOO$

Epoxy hardening propagates on top side of underfill

Problems of the Totally to Exposed Process





It is difficult to achieve voidless and solid diffusion

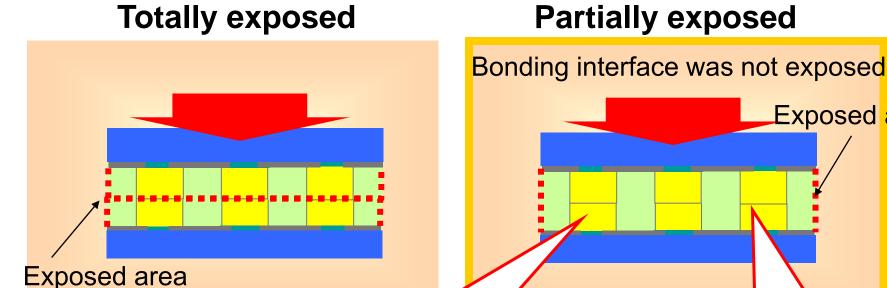
Advantage Partially Exposed Process



Exposed area

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Underfill adhesion can be maintained.



Partially exposed

Formic acid

penetration?

Compare two conditions by TEM

- N₂ atomosphere
- Formic acid vapor

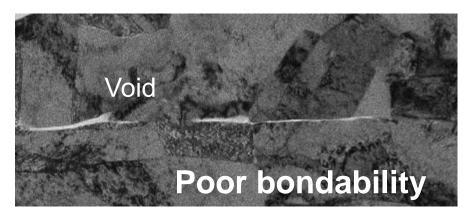
Underfill infiltration?

EDX analysis

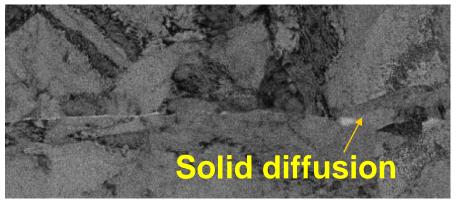


Void exists between interface under N₂ atmosphere condition.

N₂ atmosphere



Formic acid

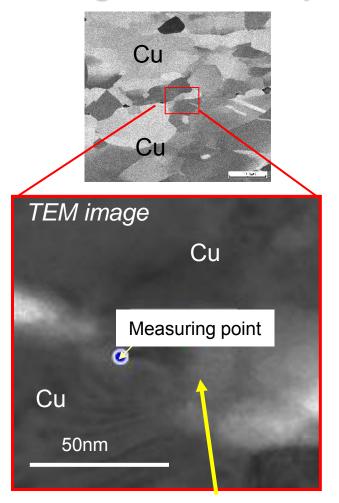


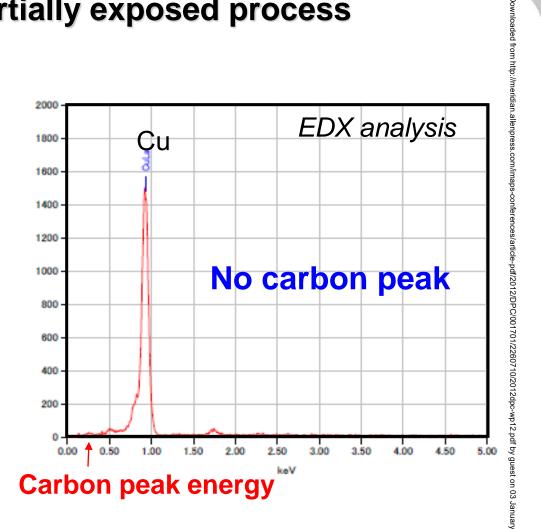
Formic acid penetrates into the bonding interface

Examination of Underline Palaring Palar



Bonding interface of partially exposed process





Cu-Cu metallic bond was confirmed

No underfill infiltration into the bonding interface

NN 1728

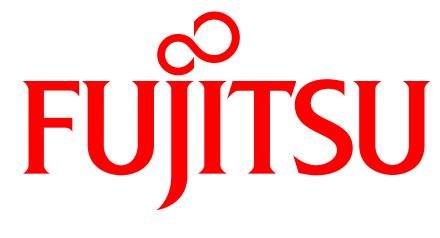
Conclusional Conference & Exhibition on Device Packaging | March 5-8, 2012 | Fountain Hills, AZ USA



- Underfill and Cu bumps can be simultaneously cut together by diamond bite with no residue on bump.
- Partially exposed process where the exposure area by formic acid is limited can achieve solid diffusion and no void formation.
- It is considered that formic acid penetrates into the bonding interface and removes oxidation layer.
- There is no underfill infiltration into the bonding interface in partially exposed process.

We have developed both underfilling and Cu-Cu solid diffusion process, which are promising methods for 3D-SIC.

Multi chip stacking process and reliability of stacked ICs are for the future work.



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